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Theoretical Study on Binary Isentropic Refrigerant Mixtures

Paper 2387

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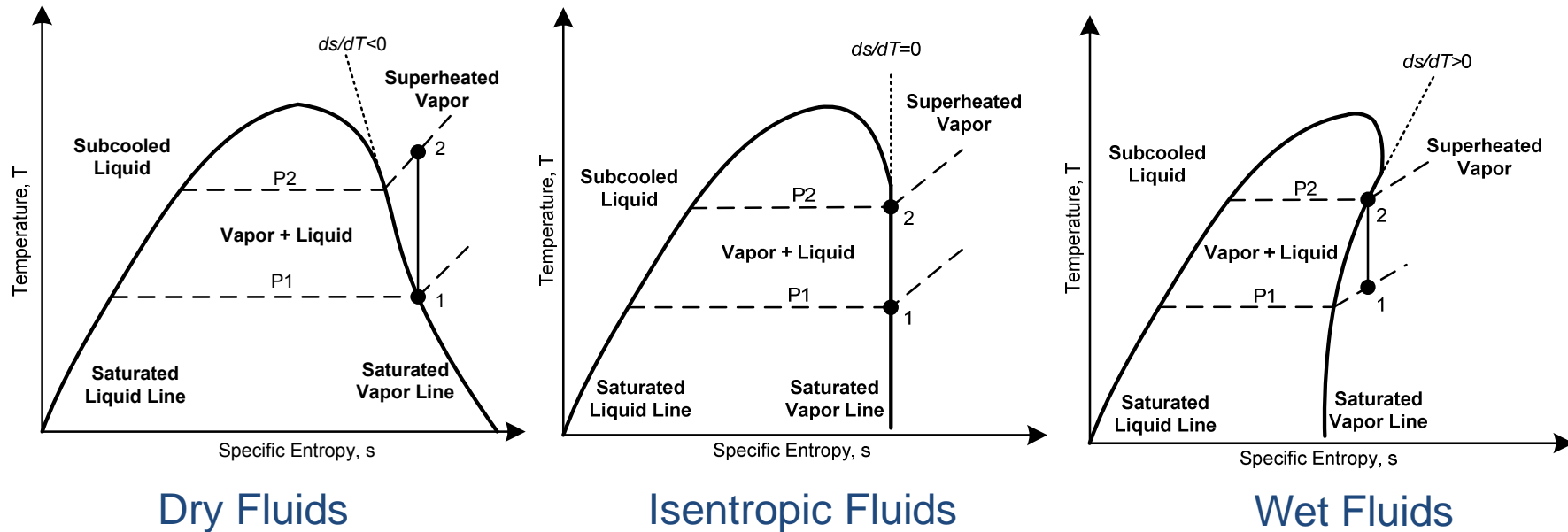
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- 🌱 Introduction
- 🌱 Superheat Parameter and Isentropic Fluids
- 🌱 Making Binary Isentropic Mixtures
- 🌱 Application of Isentropic Mixtures in Heat Pumps
- 🌱 Conclusions

Working Fluids Types in VCC

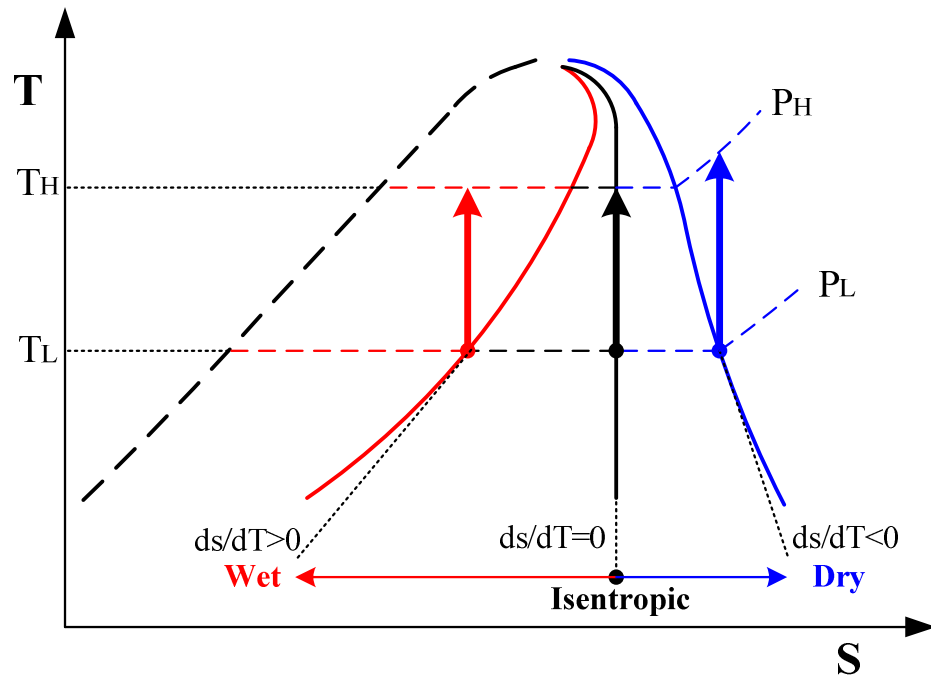


For Isentropic compression process:

- Wet fluids: saturated vapor \rightarrow two-phase
- Dry fluids: saturated vapor \rightarrow superheated vapor
- Isentropic fluids: saturated vapor \rightarrow saturated vapor

Superheat Parameter ζ

Relation between SHD and ζ



$$SHD \approx \int_{T_{evap}}^{T_{cond}} (\zeta - 1) dT \quad (1)$$

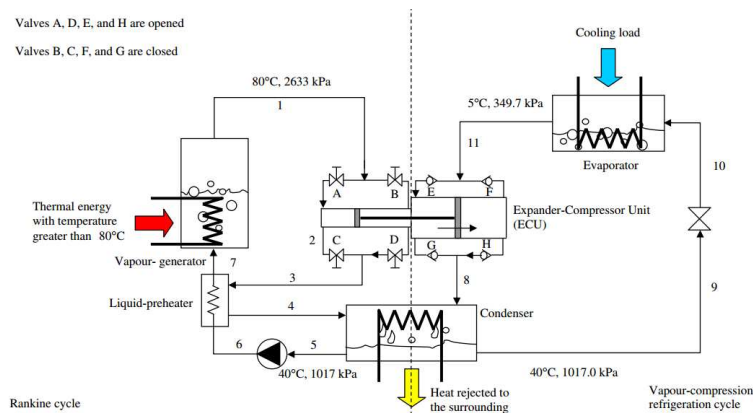
$$\zeta \equiv \beta \Delta H_{lv} V / C_P \Delta V \quad (2)$$

β : coefficient of thermal expansion (1/K)

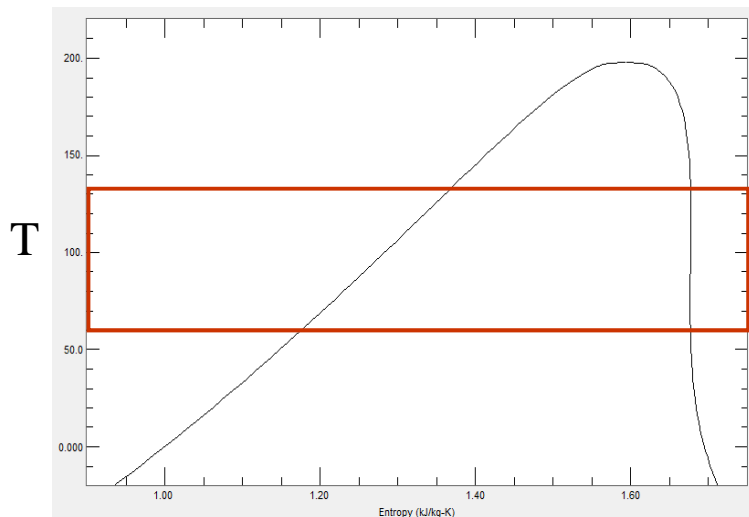
Classification based on ζ :

- $\zeta > 1$: dry refrigerants
- $\zeta < 1$: wet refrigerants
- $\zeta = 1$: isentropic refrigerants

Isentropic Refrigerant



Combined ORC-VCC system



T-s diagram for R11

- Ideal working fluids, due to:
 - Lower risk of liquid slugging
 - Lower SHD, effective utilization of HX
 - Lower compression work in theory

Restrictions

- Limited availability (e.g: R11 and R123)
- ODP restriction
- ζ varies as temperature changes



Make Isentropic mixtures

ζ of Binary Mixtures

- When $T_r = \text{const}$ and $T < T_{\text{crit}}$, ζ could be simplified as:

$$\zeta \propto P_r \left\{ \begin{array}{l} \zeta \approx \frac{\Delta H_{lv}}{C_p} \propto \frac{\varepsilon}{f(\sigma^3)} \left\{ \begin{array}{l} \Delta H_{lv} \propto \varepsilon \\ C_p \approx H^{kin} = \frac{3}{2} n K_T T \end{array} \right. \\ P_r \propto \frac{\varepsilon}{\sigma^3} \end{array} \right. \begin{array}{l} \varepsilon: \text{Intermolecular interactions} \\ \sigma^3: \text{Molecule dimensions} \\ \text{Law of Equipartition} \end{array}$$

Principle of Corresponding States

- Assume R1, R2 and their mixture RM

$$P_{crit,R1} > P_{crit,RM} > P_{crit,R2} \longrightarrow \zeta_{R1} > \zeta_{RM} > \zeta_{R2}$$

- If $\zeta_{R2} > 1$, then $\zeta_{RM} > 1$
 - If $\zeta_{R1} < 1$, then $\zeta_{RM} < 1$
 - If $\zeta_{RM} = 1$, then $\zeta_{R1} > 1 > \zeta_{R2}$
- Isentropic mixture should be made by mixing wet and dry fluids

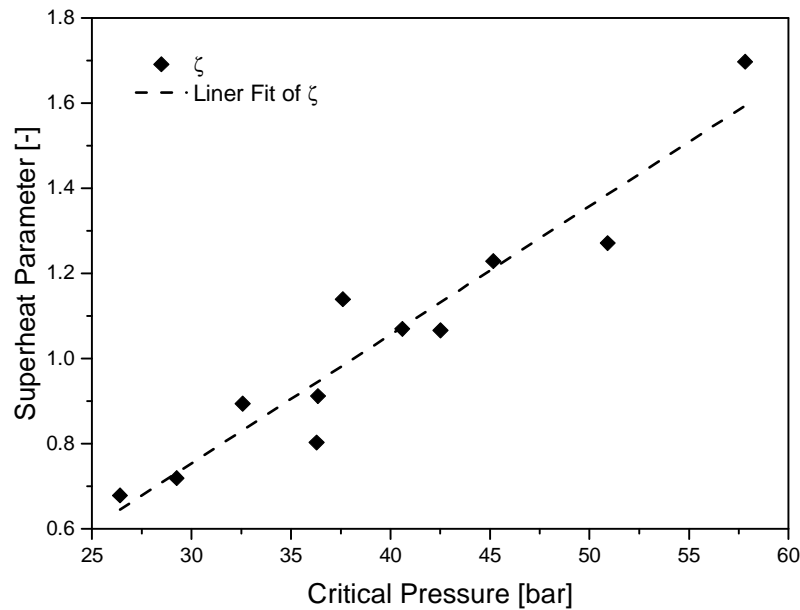
Selection of Component Refrigerants

Specified Working Conditions

HTF at Condenser		HTF at Evaporator	
Inlet T (°C)	Outlet T (°C)	Inlet T (°C)	Outlet T (°C)
45	55	-8	-13

Initial selection criteria:

- $T_{nb, \max} = -8 \text{ } ^\circ\text{C}$
- $T_{crit, \min} = 70 \text{ } ^\circ\text{C}$
- $ODP = 0$



Properties of Candidate Refrigerants

Fluid	M (g/mol)	P_{crit} (bar)	T_{crit} (°C)	T_{nb} (°C)	ζ (-)
R218	188.02	26.4	71.87	-36.79	0.6782
R227ea	170.03	29.25	101.75	-16.34	0.7187
R600a	58.12	36.29	134.66	-11.75	0.8027
R1234yf	114.04	32.57	94.70	-29.45	0.8937
R1234ze	114.04	36.36	109.37	-18.95	0.9115
R134a	102.03	40.59	101.06	-26.07	1.0694
R290	44.10	42.51	96.74	-42.11	1.0659
R143a	84.04	37.61	72.71	-47.24	1.1393
R152a	66.05	45.17	113.26	-24.02	1.2286
R161	48.06	50.91	102.15	-37.55	1.2709
R32	52.02	57.82	78.11	-51.65	1.6971

wet

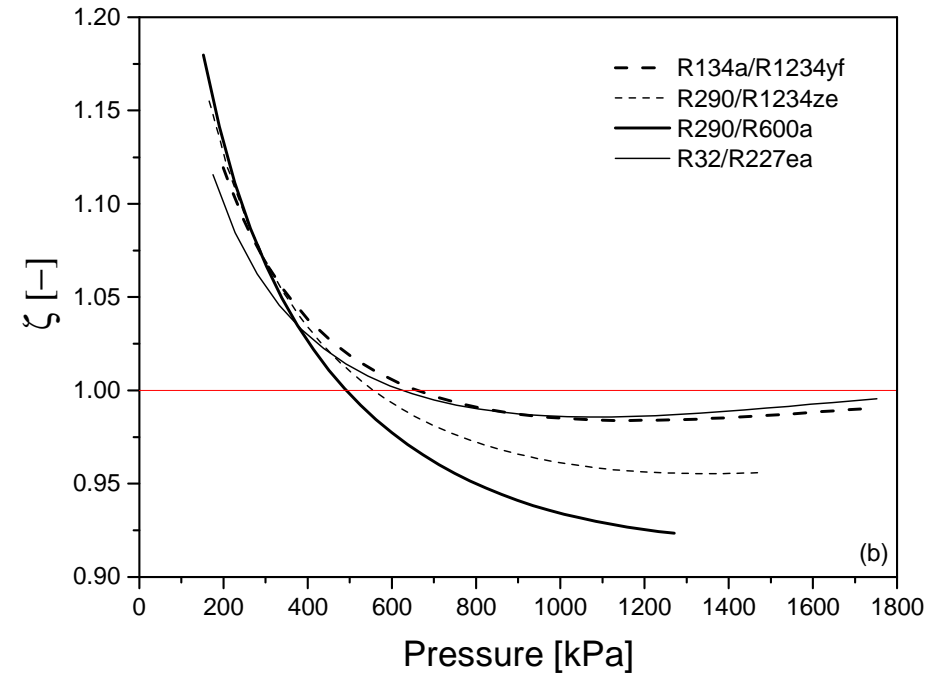
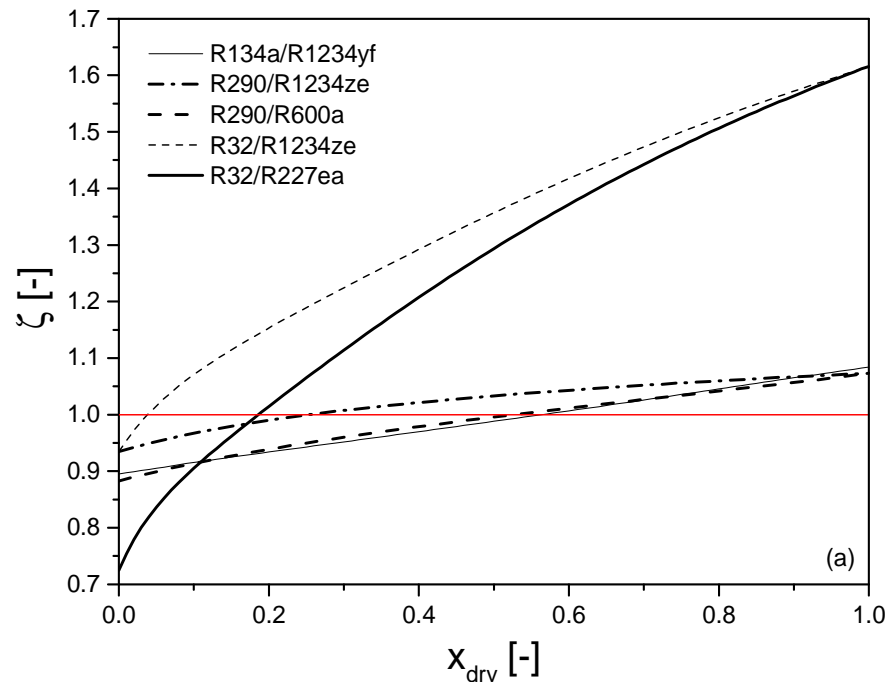
dry

Potential Binary Isentropic Mixtures

Mixture	$x_{dry}(\%)$	$\Delta T_{gl}(K)$	Mixture	$x_{dry}(\%)$	$\Delta T_{gl}(K)$	Mixture	$x_{dry}(\%)$	$\Delta T_{gl}(K)$
R32/R218	46.74	0.4	R290/R218	68.18	0	R134a/R218	74.54	4.8
R32/R1234yf	7.46	7.8	R290/R1234yf	42.01	0.8	R134a/R1234yf	56.39	0.1
R32/R1234ze	3.82	6.3	R290/R1234ze	25.38	3.3	R134a/R1234ze	43.10	0.8
R32/R227ea	18.60	12.6	R290/R227ea	62.78	0.3	R134a/R227ea	74.14	0.3
R32/R600a	2.82	19.2	R290/R600a	52.63	7.4	R134a/R600a	75.47	0.8
R143a/R218	67.86	0.2	R161/R218	36.11	6.5	R152a/R218	46.12	1.9
R143a/R1234yf	41.73	2.9	R161/R1234yf	16.08	1	R152a/R1234yf	23.87	0
R143a/R1234ze	24.81	6.4	R161/R1234ze	9.01	2.3	R152a/R1234ze	15.11	0.5
R143a/R227ea	63.45	4.4	R161/R227ea	29.54	3.3	R152a/R227ea	40.51	0.5
R143a/R600a	60.04	15.4	R161/R600a	20.80	10.6	R152a/R600a	35.93	6.7

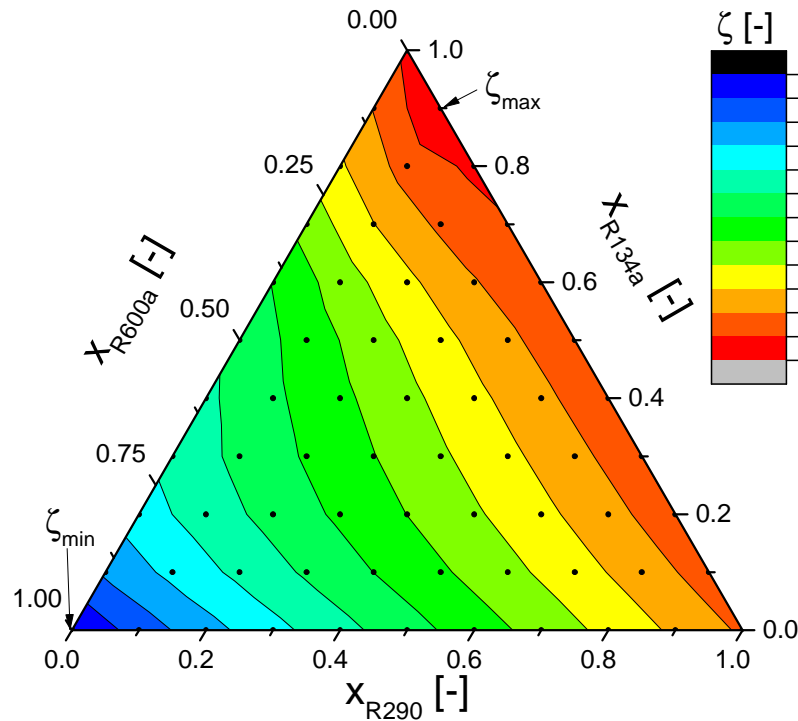
- 30 binary isentropic mixtures are obtained.
- Temperature glide is calculated based on condensing pressure.
- 5 typical mixtures include HC/HC, HC/HFC, and HFC/HFC.

Influence of Composition and Pressure on ζ

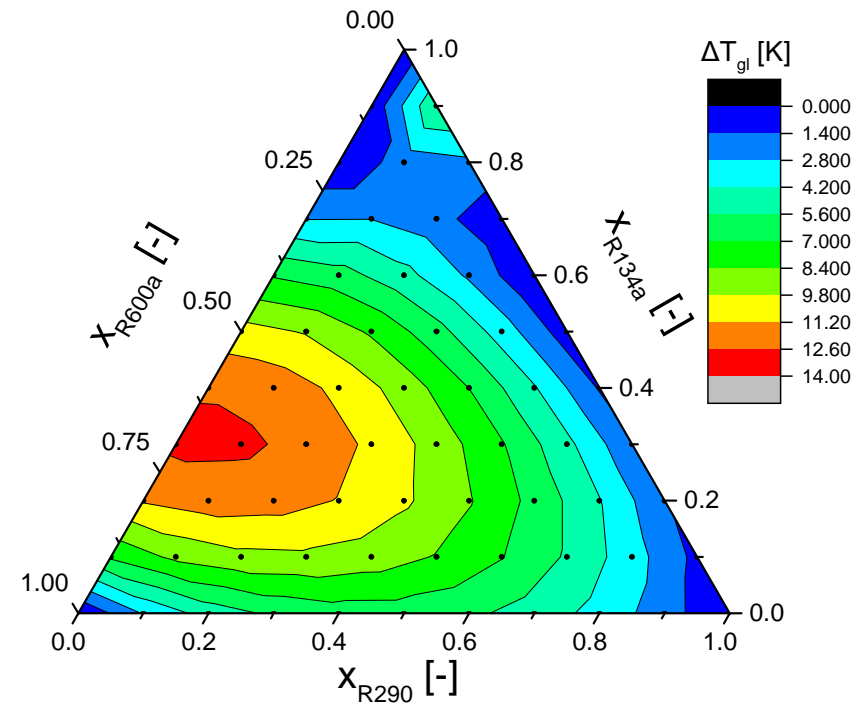


- ζ increases from <1 to >1 with x_{dry} .
- ζ decreases from >1 to <1 with pressure.
- How to reduce the influence of pressure for R290/R600a?

Method I: Adding a Third Component



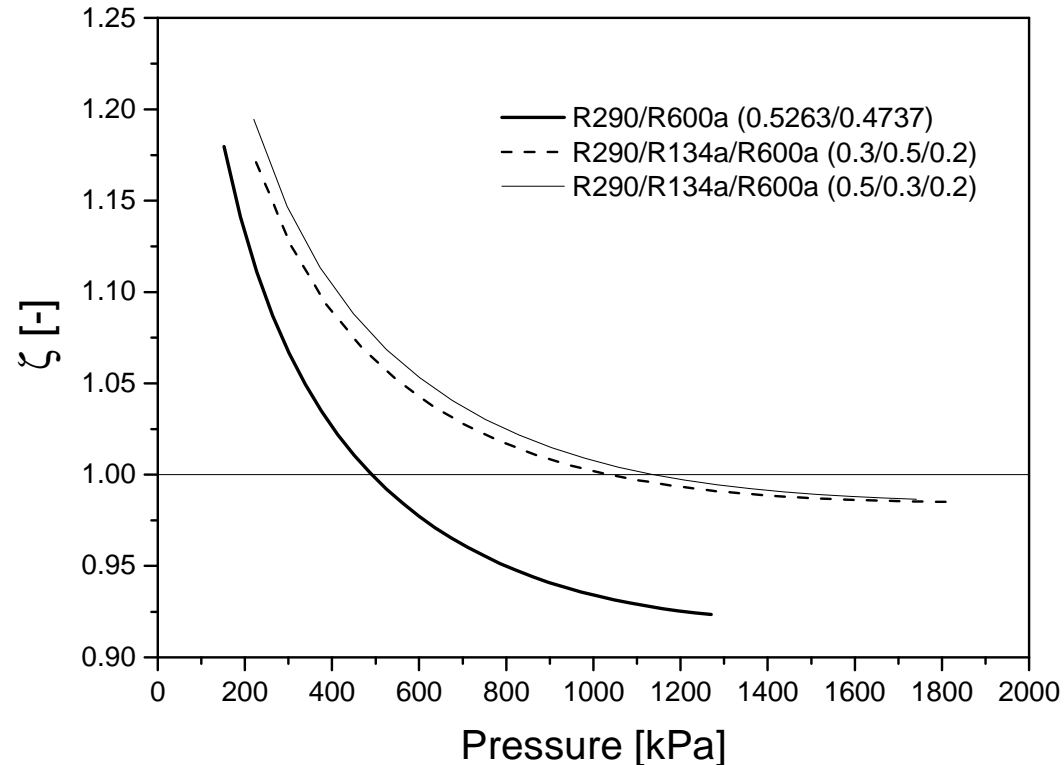
ζ -x relation in colored ternary map



ΔT_{gl} -x relation in colored ternary map

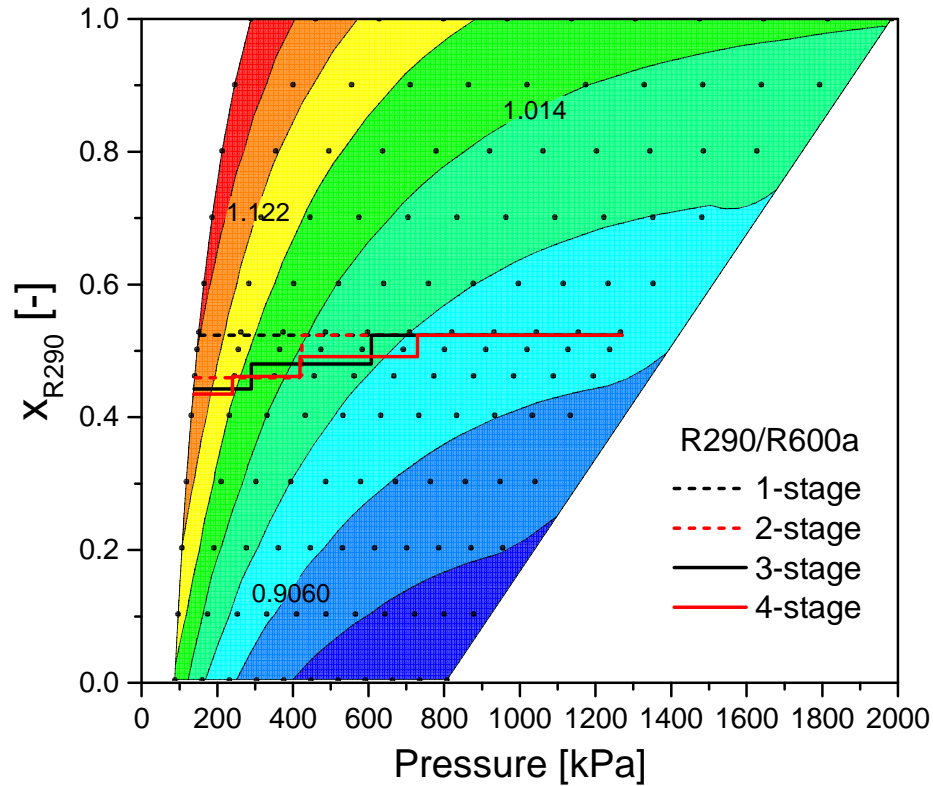
- T_{crit} : R290 < R134a < R600a, P_{crit} : R290 > R134a > R600a
- Banding distribution of ζ : $x_{R600a} \approx 0.2$, $\zeta = 1$
- Round distribution of ΔT_{gl} : $0.1 < x_{R134a} < 0.6$, $\Delta T_{gl} > 5 \text{ } ^\circ\text{C}$

Method I: Adding a Third Component



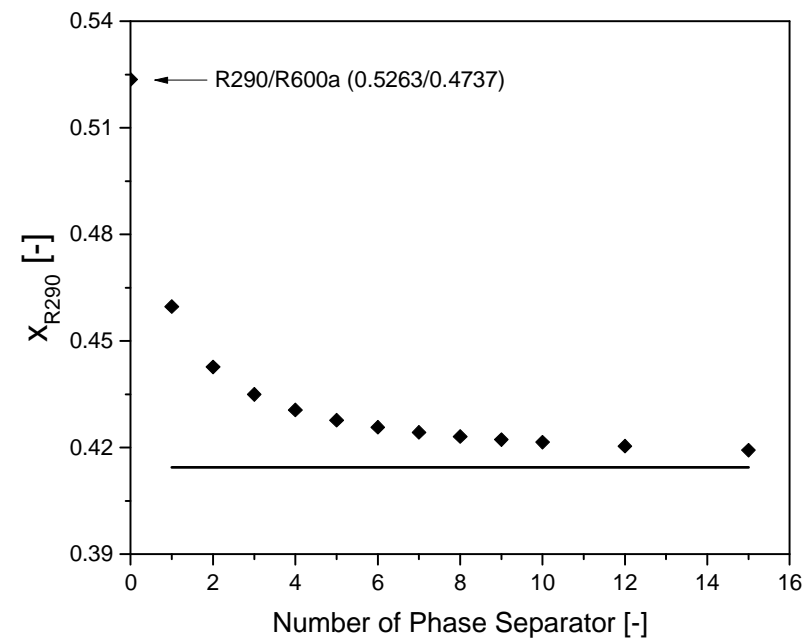
Adding third component could not reduce ζ within low pressure range.

Method II: Controlling Circulation Composition



Counter map of ζ versus composition and pressure

- ζ decreases as $P_{\text{evap}} \rightarrow P_{\text{cond}}$
- Multistage compression with refrigerant injection



Thermodynamic Model and Fluids

Model Assumptions:

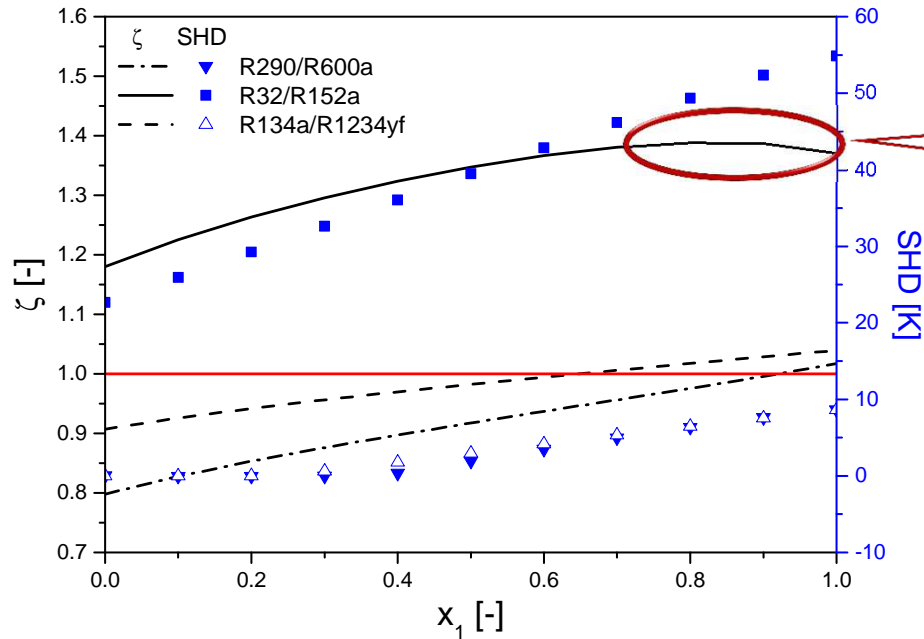
- Counter-flow type heat exchangers
- Constant overall heat transfer coefficient, U
- Constant total heat transfer conductance per unit heating capacity, as follows:

$$0.36 \leftarrow \frac{UA_{total}}{Q_c} = \frac{Q_e}{Q_c} \frac{1}{\Delta T_{lm,e}} + \frac{1}{\Delta T_{lm,c}}$$

Fluids:

Fluid Type	Zeotropic	Azeotropic
Isentropic	R290/R600a	R134a/R1234yf
Dry	R32/R152a	

Superheat Degree and ζ

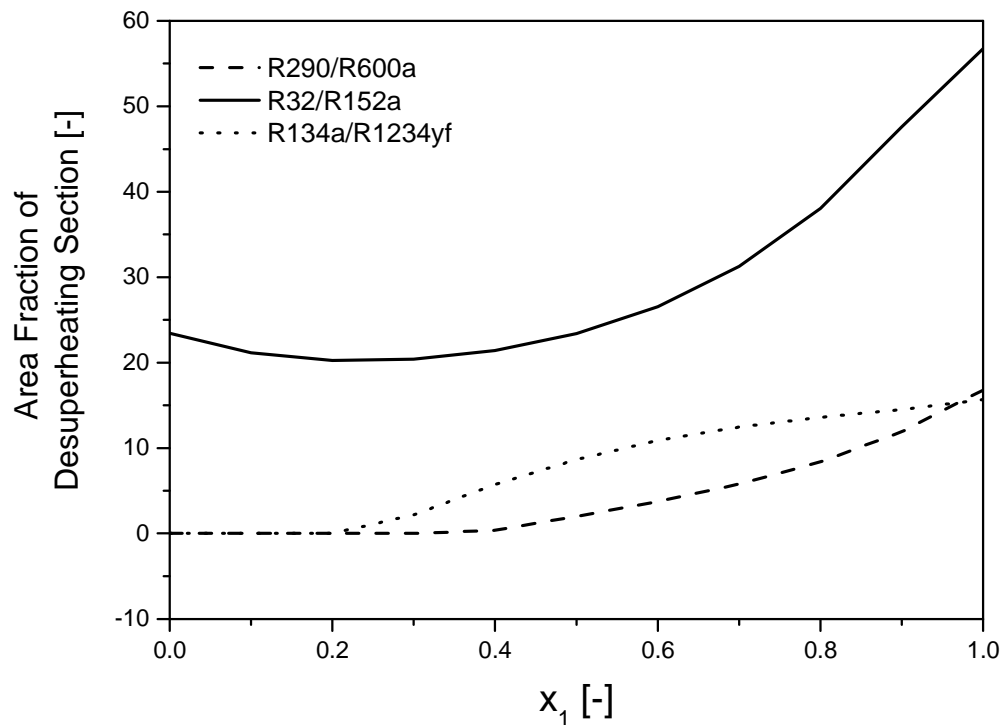


R32 has much lower critical T (78.11 °C).

- ζ is calculated based on saturated T under condensing P
- Mixture composition: mass fraction of the first component fluid (x_1)

- Generally, SHD increases with ζ .
- Dry mixture shows much higher SHD (max=55 °C) than isentropic mixtures.

Heat Transfer Area for Desuperheating



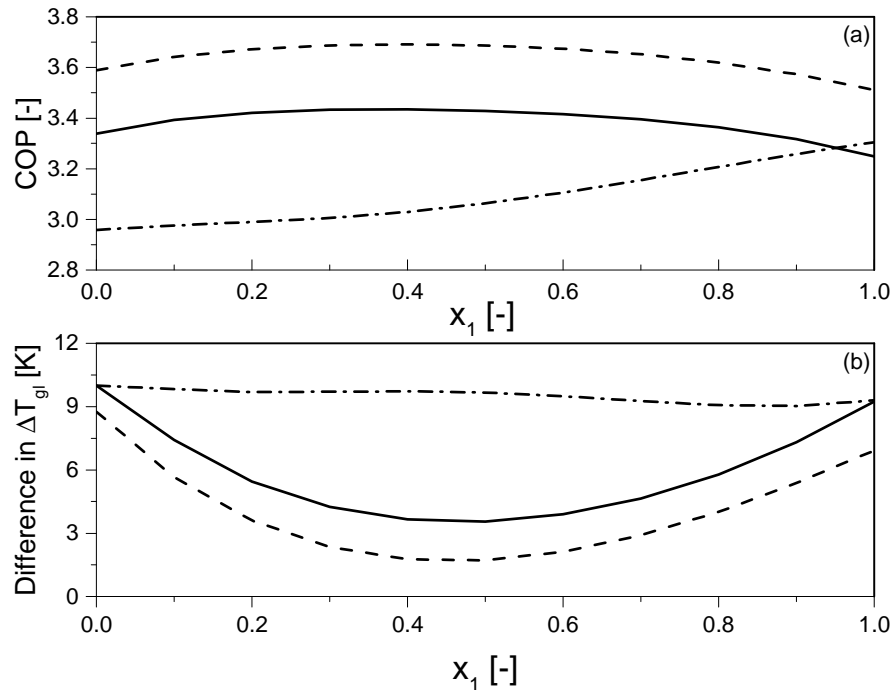
The fraction of condenser area for desuperheating could be estimated by:

$$\frac{A_{desup}}{A_{cond}} = f_{desup} \left(\frac{\Delta T_{lm,c}}{\Delta T_{lm,desup}} \right) \times 100$$

- For dry mixture R32/R152a, maximum fraction of A_{desup} is 55%.

Heating COP

Maximum COP: R32/R152a > R290/R600a > R134a/R1234yf



- R32/R152a: maximum COP=3.69 ($x_1=0.4$), minimum $\Delta T_{gl}=1.7$ K ($x_1=0.5$)
- R290/R600a: maximum COP=3.43 ($x_1=0.4$), minimum $\Delta T_{gl}=3.6$ K ($x_1=0.5$)
- The ever-increasing SHD hinders the better match of temperature glides.

Solid line: R290/R600a; Dash line: R32/R152a;
Dash dot line: R134a/R1234yf

Conclusions

- **Introduce the concept of isentropic mixtures.**
- **Mixing dry and wet fluids could lead to binary isentropic mixtures.**
- **By using isentropic mixtures, SHD as well as condenser area for desuperheating could be reduced.**

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Thank You!